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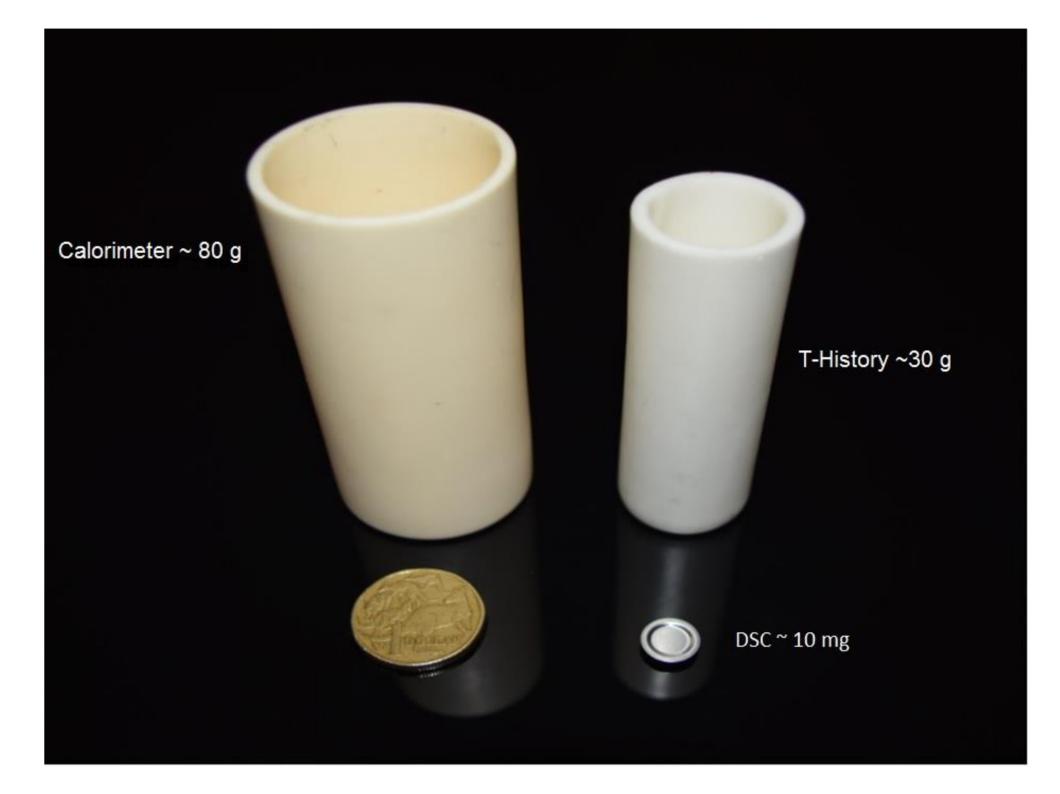
Evaluation of the Thermophysical Properties of High Temperature Phase Change Materials for Thermal Energy Storage

Australian Concentrating Solar Thermal Symposium and Workshop - 2016 Ehsan Omaraa¹, Wasim Saman¹, Frank Bruno¹ and Ming Liu¹

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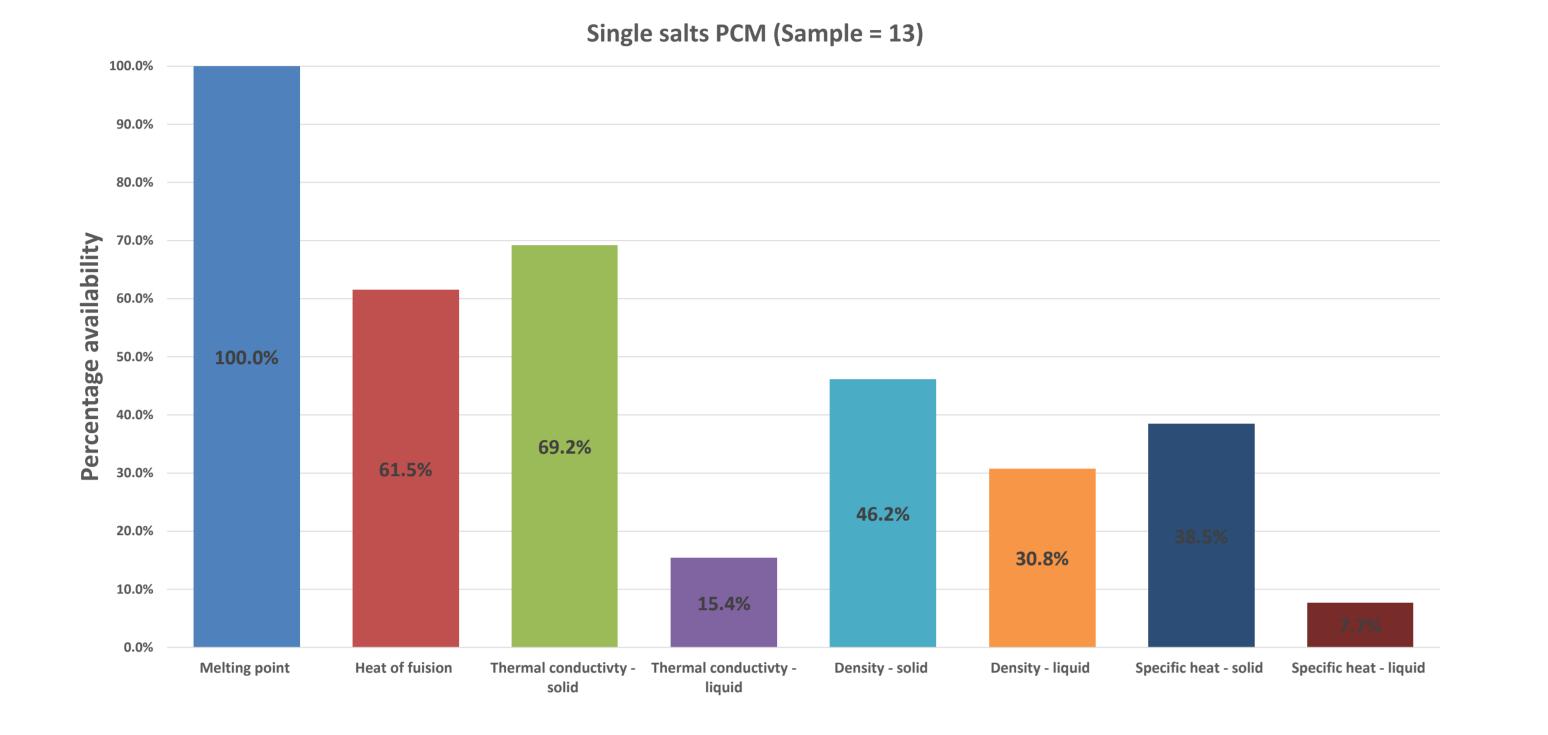
Background

- There is limited data on the thermophysical properties (melting/freezing point, enthalpy, specific heat capacity) of phase change materials between 300 °C to 700 °C.
- Methods of measurement use small test samples (1–10mg), and the data is limited to homogeneous materials.

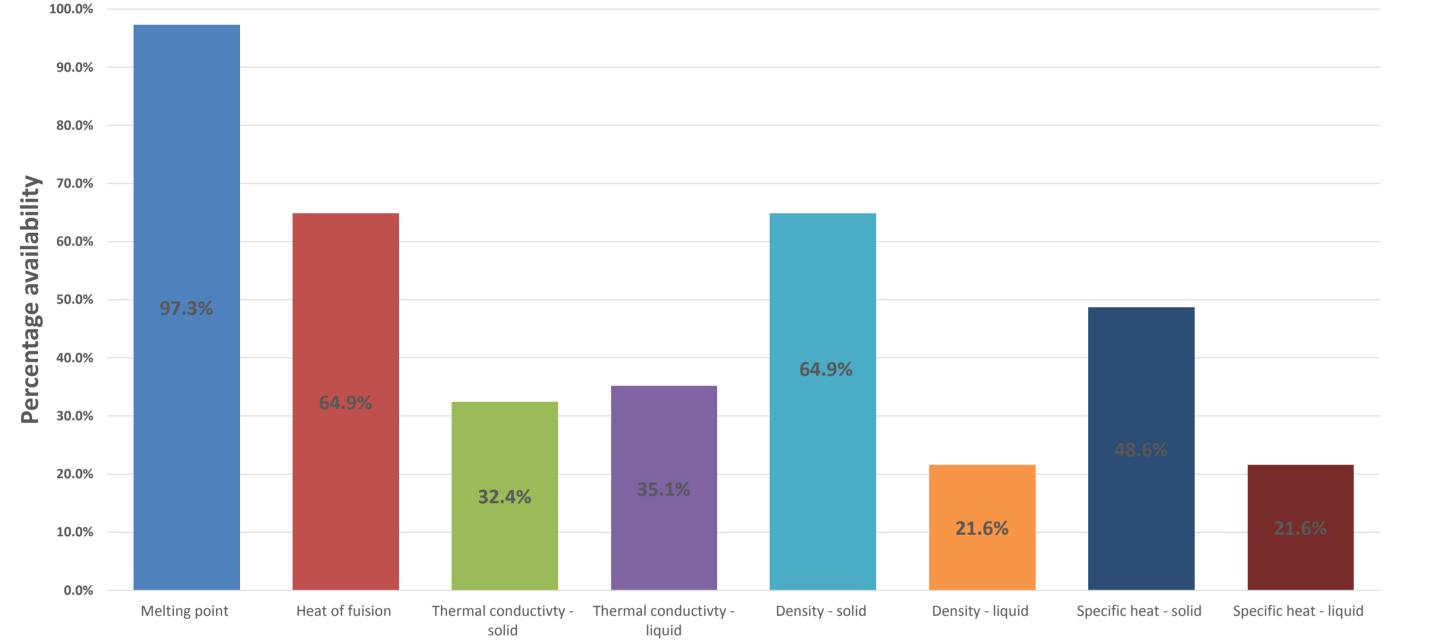


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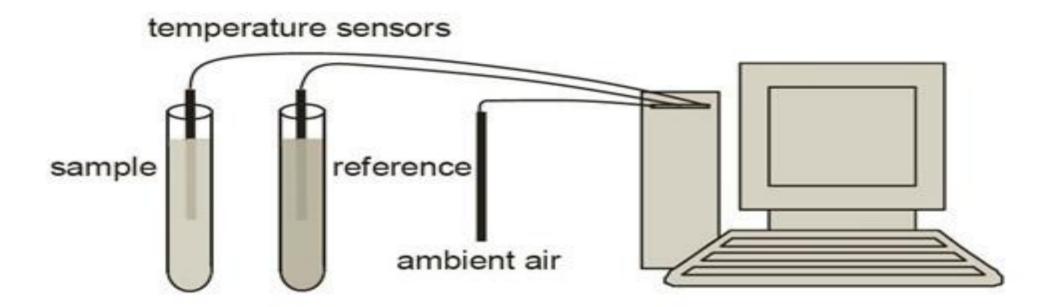


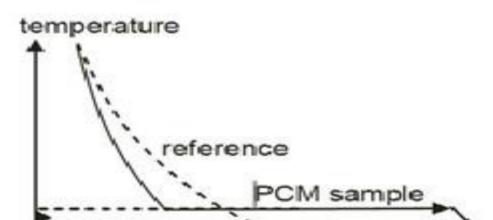
PCM with more than 1 salt (Sample = 37)





Test Bench





phase change ambient time

Figure 3: A schematic diagram of T- history (Mehling & Cabeza 2008).

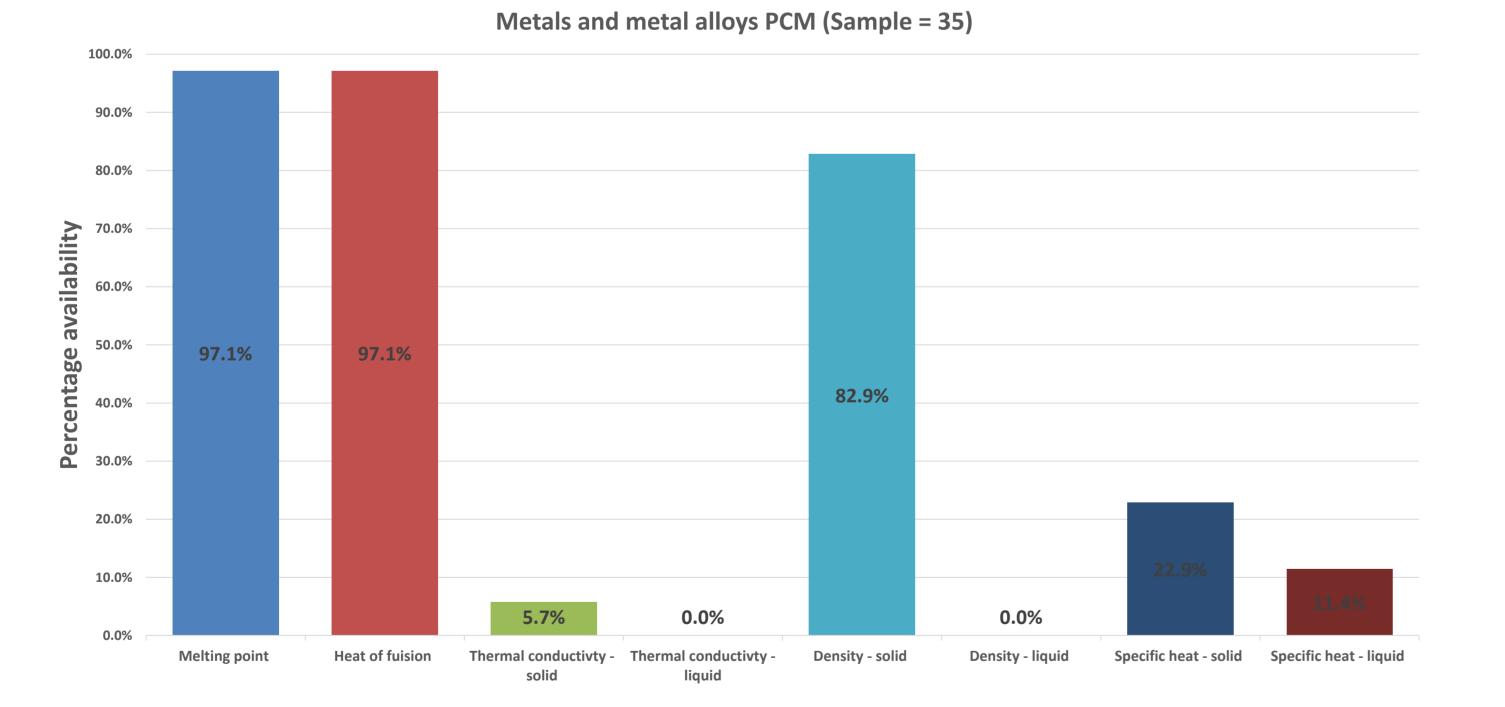
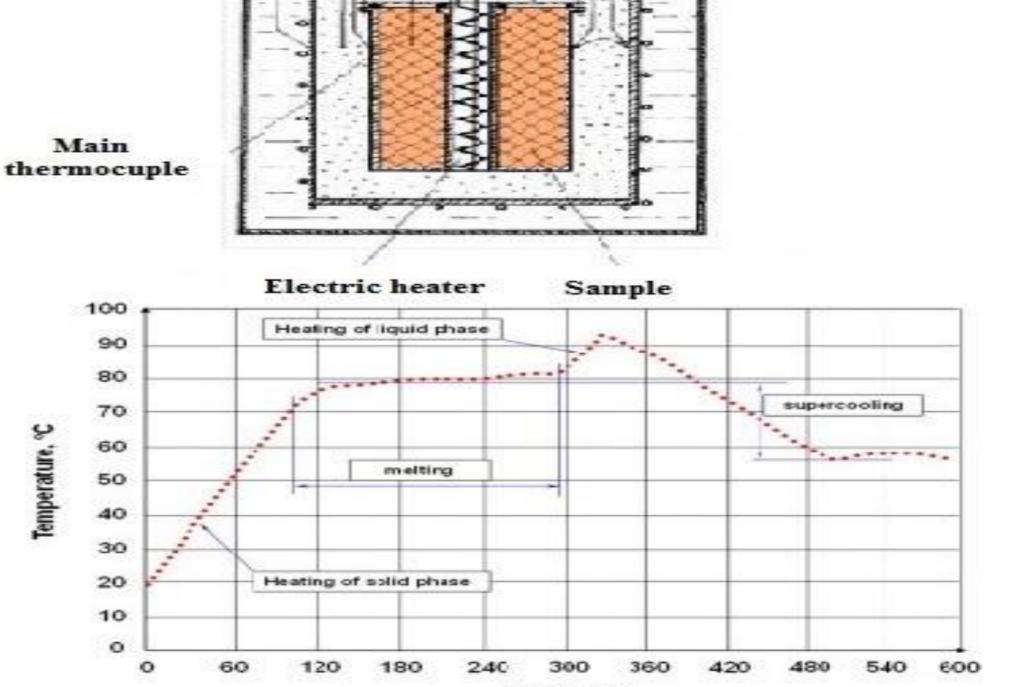


Figure 1: Statistics of the availability of thermophysical properties from existing literature (Omaraa et al. 2015)

Motivation



Time, min

- PCMs are heterogeneous materials
- Properties of materials may vary with volume (due to subcooling, phase separation)
- Conventional instrument is a differential scanning calorimeter, which measures volumes of around 1 to 10 milligrams
- Other methods such as T-history and calorimeter measure in grams.

Figure 4: A schematic of the calorimeter (Domanski, Jaworski & Rebow 1995).

Conclusion

The calorimeter and temperature history (T-history) method may be simpler, less expensive and more accurate than current methods.

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